

Enhancement of performance of TiO₂/Cu₂O solar cells

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The TiO₂/Cu₂O heterojunction solar cell was investigated in the study given by using the Solar Cell Capacitance Simulator (SCAPS). It has been thoroughly investigated how different layer factors, such as thickness, defect density and acceptor density affect cell performance. The efficiencies (η) of solar cell were shown to be influenced by the thickness of the absorber (Cu₂O) and buffer (TiO₂) layers using numerical analysis. TiO₂/Cu₂O optimized solar cell design demonstrated potential efficiency of about 23%. We then looked at how temperature affected the photovoltaic efficiency.

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1. Introduction

The most encouraging semiconductors for the photovoltaic (PV) applications are the metal oxide semiconductors (MO). Because many MOs are non-toxic, chemically stable, and rich which permits material deposition under surrounding conditions [1]. MOs are already widely used in a variety of current commercial applications as active or passive components, such as the active channel layer in the transistors that make up the active matrix of displays [2] or the transparent conducting front electrodes and electron or hole transport layers in solar cells [3]. Due to its potential to lower costs by utilizing inexpensive materials and production techniques, heterojunctions fully based on MOs, also known as all-oxide photovoltaic cells, have recently attracted a great deal of attention [4-6].

TiO₂ is a well-known photo-catalytic material and an n-type semiconductor with a wide band-gap energy of 3.2 eV [7]. A dye-sensitized solar cell (DSSC), also known as a Grätzel cell, was created by Grätzel et al. and can increase efficiency by 11% when ruthenium dye is used [8]. However, there are issues with dye desorption, leakage, packing, and long-term stability with the Grätzel cell. The cells have undergone a number of improvements. One possibility is to change the cell from a liquid electrolyte to a solid-state cell by employing a p-type semiconductor as a hole conductor. In an effort to convert cells into solid-state DSSCs, a range of p-type semiconducting materials, including Spiro-OMeTAD [9] and copper (I) iodide (CuI) [10-11], have recently been used. However, interaction between the dye monolayer and the p-type material is crucial in solid-state DSSCs. Previous attempts to address these difficulties have failed since only partial filling of the TiO₂ pores with a p-type material was achieved, especially when thicker films were used.

One way to overcome the problems with solid-state DSSCs is to use ETA (extremely thin absorber) solar cells

or quantum dot (QD) sensitized solar cells, which are conceptually similar to solid-state DSSCs [12–15]. In those solar cells, a small band-gap p-type semiconductor, such as CuInS₂, CdTe and SnS replaces the molecular dye in the DSSC and works as a photon absorber in the cells [16–18]. The semiconductor typically covers the n-type semiconductor film, which is usually TiO₂. The structure of the ETA and QD solar cells has the advantage of enhanced light harvesting due to surface enlargement and multiple scattering [7,8]. The TiO₂/CdTe cells fabricated by Ernst et al. [16] exhibited an open-circuit voltage of 0.67 V and a short-circuit current of 8.9 mA/cm² under 100 mW/cm² of simulated sunlight. Nanu et al. [17] generated TiO₂/CuInS₂ solar cells using an atomic layer chemical vapor deposition method (ALCVD) with 4% solar energy efficiency.

The most well studied system is heterojunction ZnO/Cu₂O cells [19], where the wide bandgap ZnO serves as a window layer and the Cu₂O has a bandgap at around 2 eV in the visible spectrum of the sun [20]. The impressive light to electric power conversion efficiencies have been recorded for cells with a ZnO window layer up to 4%, [21] while 5% were attained with a Ga₂O₃ layer [22], despite the Cu₂O bandgap not being appropriate for sunlight (AM1.5G). Recently, a Ga₂O₃/Cu₂O heterojunction cell was reported to have an open circuit voltage of up to 1.2 V [23]. The Cu₂O bandgap is virtually optimal for multi-junction tandem cells with three or more connections [24]. Cu₂O is also a very desirable absorber for semi-transparent photovoltaics. Cu₂O thin films have been used in optoelectronic devices such thin film transistors in addition to solar cells [25].

Solar Cell Capacitance Program (SCAPS-1D) numerical simulations of solar cells have been reported. Planar thin-film CdS/CdTe solar cells and CdS/CdTe nanowires were simulated by Anwar et al. [26]. They showed that the CdS nanowires increase the CdS/CdTe solar cells' efficiency by about 3%. It has also been examined how temperature, interface state density, and

interface surface recombination velocity affect CdS/CdTe nanowire solar cells. For thin film solar cells, Decock et al. [27] simulated multivalent defects with up to five different charge states and examined the properties of current density-voltage (J-V) and capacitance-frequency (C-f). They came to the conclusion that flaws affect the simulation's outcomes [27]. For polycrystalline CdTe and CuInSe films, Niemegeers et al. [28] created a numerical simulation device to determine the J-V and C-f properties. Then, comparisons between numerical simulations and actual measurements were made. For a trustworthy interpretation of doping profile measurements, numerical simulations can be utilized. The concordance between simulations and measurements was examined by Burgelman et al. [29] who modelled CdTe and Cu(In, Ga)Se₂ solar cells and presented the outcomes of J-V, capacitance-voltage (C-V), and C-f simulations. n-TiO₂/p-CuO and n-TiO₂/p-Cu₂O heterojunction solar cells were simulated by Sawicka-Chudy et al. [30] who also examined the impact of layer thickness and defect density in buffer and absorber thin films on cell performance. On the basis of the numerical simulations, Sawicka-Chudy et al. [31] also provided numerical research of J-V curves for n-TiO₂/p-Cu₂O solar cells and confirmed the potential usage of the n-TiO₂/p-Cu₂O structure.

However, there are still problems that need to be solved, such as the optimum work configuration, ideal material qualities, desired, and boundary work conditions for high cell performance. As a result, the authors made the decision to continue their research on structural analysis in order to obtain thorough results.

2. Methods

The schematic diagram of the proposed device is shown in (Fig. 1). The solar cell structure consists of three different layers: TiO₂ (buffer), Cu₂O monolayer (absorber) and the substrate. The proposed structure was simulated with the help of SCAPS-1D. The carrier transport and recombination process at the interface was investigated using Shockley-Read-Hall (SRH) interface approach.

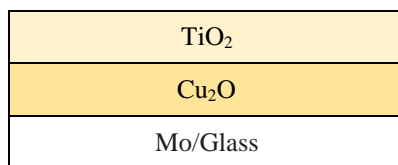


Fig. 1. Structure of the solar cell simulated by using SCAPS-1D

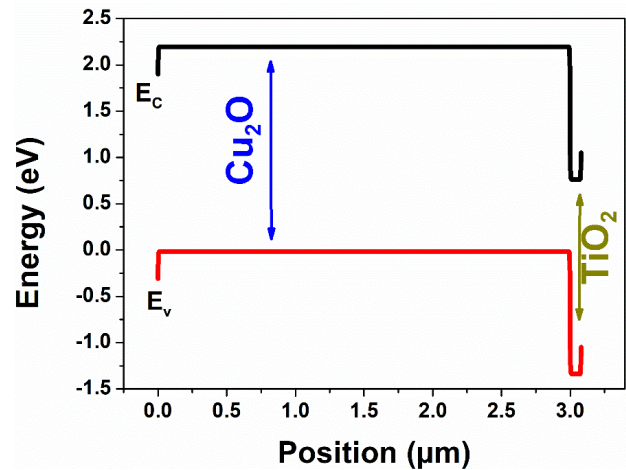


Fig. 2. Band gap diagram

Fig. 2 describes the energy diagram for the proposed structure of solar cell. It infers the electron affinity and band gap of the simulated solar cell. The obtained bandgap values of Cu₂O and TiO₂ are 2.15 eV respectively which are in the range. The band gap diagram was drawn using parameters as: solar spectrum AM1.5, $P = 100 \text{ mW/cm}^2$, and $T = 300 \text{ K}$. Other basic numerical parameters used in simulation of proposed solar cell layers are given in (Table 1).

Table 1. Materials and interface Parameters used in SCAPS-1D simulation

Parameters	n-TiO ₂	p-Cu ₂ O
E _g (eV)	2.26	2.17
Electron affinity (eV)	4.2	3.2
Dielectric permittivity	10	7.11
CB effective density of states (cm ⁻³)	2.2 x 10 ¹⁸	2.2 x 10 ¹⁸
CB effective density of states (cm ⁻³)	1.8 x 10 ¹⁹	1.8 x 10 ¹⁹
μ _n (cm ² /V.s)	100	200
μ _p (cm ² /V.s)	25	80
Parameters (Unit)	TiO ₂ /Cu ₂ O interface	
Type of defect	Neutral	
Electrons capture cross-section (cm ²)	1 x 10 ¹⁹	
Holes capture cross-section (cm ²)	1 x 10 ¹⁹	
Reference for defect energy level (eV)	Above the highest eV	
Energy with respect to reference (eV)	0.6	
Total defect density (cm ⁻²)	1 x 10 ¹¹	

3. Results and discussion

3.1. Effect of thickness

The solar cell's most crucial component is the absorber layer, which is where incident photons are absorbed, and excess carriers are produced. The n-buffer layer's primary

function is to join the p-absorber layer at a p-n junction. To reduce the PV device's series resistance, the buffer layer should be as thin as possible [30].

By varying the thickness from 1.0 to 9.0 nm (Cu₂O) and from 0.02 to 0.08 nm (TiO₂) without introducing additional defects, the effect of absorber and buffer layer thickness on cell efficiency (η) is simulated in this study (Figs. 3, 4).

Cell efficiency decreases linearly in relation to TiO₂ layer thickness in the TiO₂/Cu₂O structure. As a result, the TiO₂ buffer layer's optimal thickness for the current study has been set at 0.08 nm for TiO₂/Cu₂O. This indicates a cell efficiency of 22.65 percent.

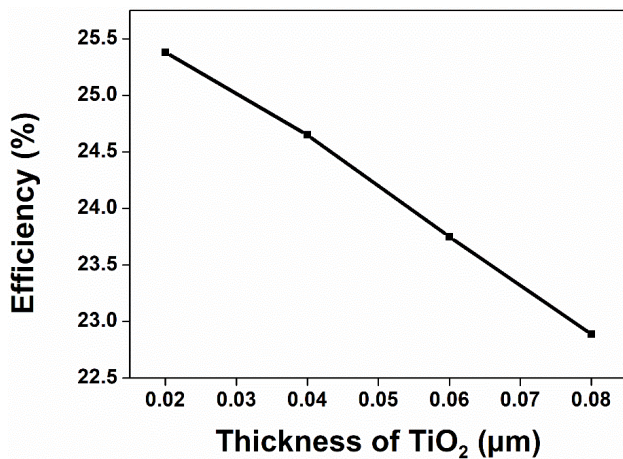


Fig. 3. Thickness of TiO₂ layer effect on the efficiency

It is essential to note that the curves in Figs. 3 and 4 have distinct trends. As the thickness of the Cu₂O layers increases, the efficiency of the solar cell initially increases and eventually reaches saturation at higher values. Nonetheless, as the thickness of the TiO₂ layers increases, so does the efficiency of the solar cell. The layers of Cu₂O can take in more photons and generate more current as the layer's thickness increases, which are mostly due to the thicker absorber layer's increased ability to absorb incident light [31]. Due to the limited diffusion length of the carrier, an increase in absorber thickness increases the likelihood of SRH (Shockley-Read-Hall) recombination, which results in this efficiency. As a result, the optimal thickness for the Cu₂O absorber was set at 3.0 nm for the current investigation. The cell is 22.65 percent efficient.

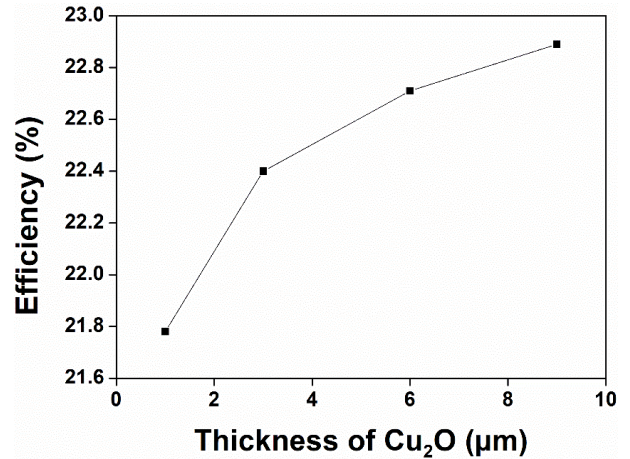


Fig. 4. Thickness of Cu₂O layer effect on the efficiency

3.2. Effect of defect density in n-TiO₂ and acceptor density in p-Cu₂O

The efficiency of proposed solar cell was also investigated in context of natural defects density in n-TiO₂ buffer layer. (Fig. 5) shows the variation in efficiency for density in the range from 10¹⁷ to 10²⁰ cm⁻³, however the thickness of n-TiO₂ buffer layer was kept as 50 nm. The observed improvement in efficiency can be justified by the introduction of additional carrier recombination centers within the layer [32].

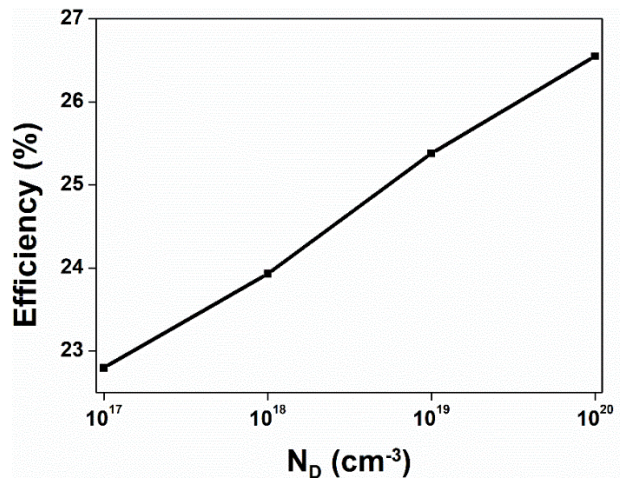


Fig. 5. N_D of TiO₂ layer effect on the efficiency

The efficiency of Cu₂O based solar cell was observed to increase from 24.6 to 25.7 % as the Cu₂O acceptor density varies from 10¹⁷ to 10²⁰ cm⁻³ leading to the availability of added charge carriers for conduction within the cell as shown in (Fig. 6).

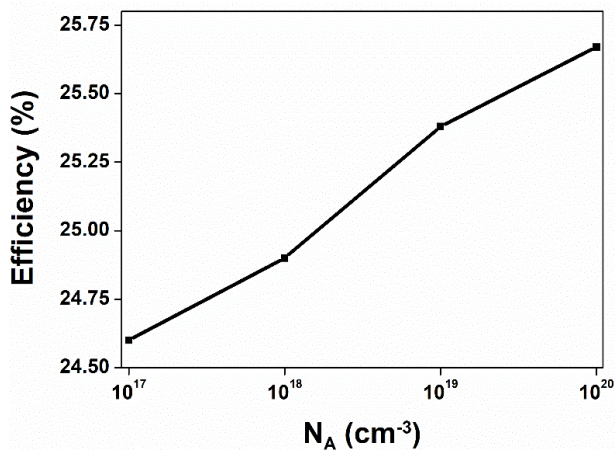


Fig. 6. N_A of Cu_2O layer effect on the efficiency

3.3. Effect of temperature on efficiency

The temperature plays a significant role in achieving the highest efficiencies of the solar cell. The temperature range is varied from $0^{\circ}C$ - $40^{\circ}C$. It can be inferred from the (Fig. 7) that with increase in temperature, the efficiencies of the solar cells decrease. This is due to increased recombination rates due to increased carrier concentration [33].

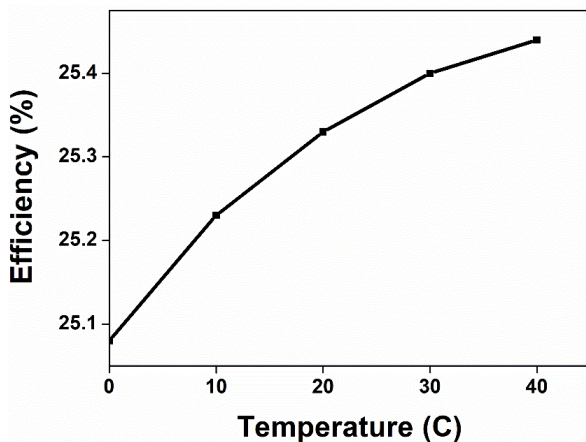


Fig. 7. Temperature effect on the efficiency

4. Conclusion

Using the SCAPS-1D solar cell simulator, the TiO_2/Cu_2O structure was examined in this work. A sophisticated simulation model of TiO_2/Cu_2O structure was finished by the analysis of the literature. Different parameters, including thickness (Cu_2O and TiO_2), defect density, acceptor density and temperature were tuned for the solar cells.

We have shown, via modelling, that changes in the absorber layer's thickness have a significant impact on conversion efficiency. The optimal value of the cell thickness was found to be 3.0 nm for the Cu_2O and 0.03 nm for the TiO_2 layer after the photovoltaic parameters were calculated.

We can also state that the defect density has a substantial effect on the performance of TiO_2/Cu_2O solar cell. With efficiency of 22.65% for the TiO_2/Cu_2O structure it demonstrated the high potential for proposed structure optimization. The technology trials that result in the development of the optimization cell will use the acquired information in practice.

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